

Nighttime Lagrangian Measurements of Aerosols and Oxidants: Homogeneous and Heterogeneous Chemical Processing of NO_x

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1. INTRODUCTION

Although heterogeneous chemical processes involving trace gases and aerosols are poorly understood they are expected to play an important role at night. ASP's 2002 Nighttime Aerosol Oxidant Plume Experiment (NAOPEX) was designed to study the chemical evolution and interaction of ambient aerosols and trace gases in the absence of photochemistry. Quasi-Lagrangian measurements were made with DOE's G1 aircraft in the nocturnal residual layer downwind of Boston, MA.

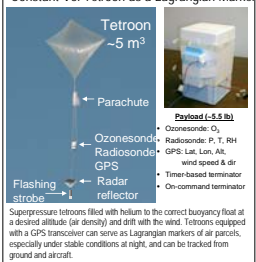
2. G1 AIRCRAFT MEASUREMENTS



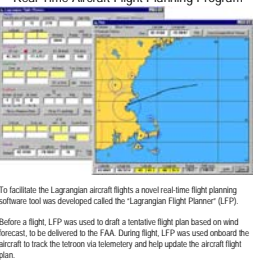
Measurement	Instrument	Groups
Aerosol Size Distribution	FSSP, PCASP, CPC, TSEMS	PNNL, Brechtel
Aerosol Composition	AMS, PILS, Single Particle Analyzer	Aerodyne, BNL, PNNL
Aerosol Optical Properties	Integrating Nephelometer, Radiance PSAP	PNNL
Inorganic Gases	O ₃ , CO, SO ₂ , NO, NO ₂ , NO _y	PNNL & BNL
VOC	Canister sampling system	York Univ.
Meteorology	T, P, Dew Pt, Vector Winds (Gust Probe)	PNNL

3. LAGRANGIAN FLIGHT PLANNING TOOLS

Constant-Vol Tetraon as a Lagrangian Marker

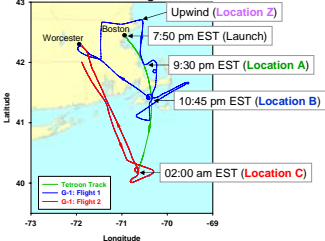


Real-Time Aircraft Flight Planning Program



4. JULY 30-31 EPISODE

G1 and Tetraon Flight Tracks

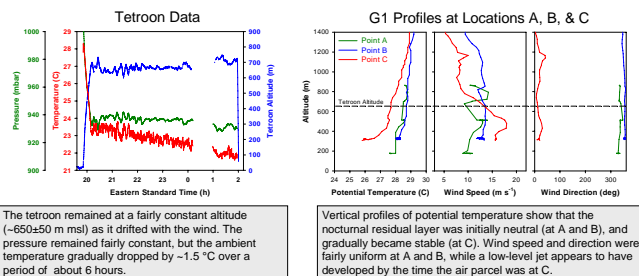


Lagrangian measurements were made within a column of air, which passed over the Salem coal-fired power plant around 7:35 pm on July 30, and continued to move south through the night. (Sunset ~7:00 pm EST)

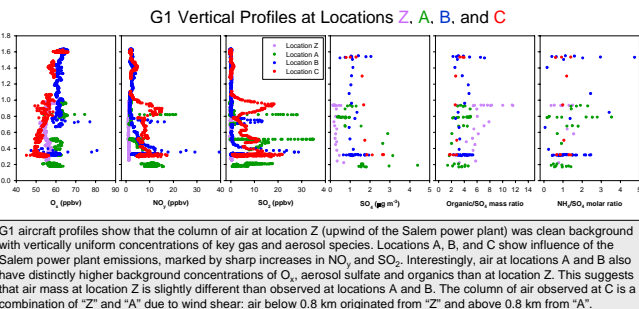
The G1 aircraft made four vertical profiles in the vicinity of the tetraon at different times, marked as locations Z, A, B, and C in the adjacent figure.

Time evolution of the meteorological variables and chemical species are described next.

5. METEOROLOGICAL OBSERVATIONS



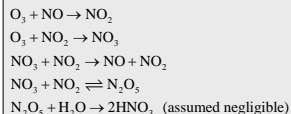
6. QUASI-LAGRANGIAN GAS-AEROSOL OBSERVATIONS



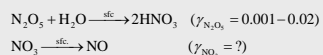
7. MODEL ANALYSIS

Nighttime Chemical Processing of NO_x

Gas-phase Reactions



Heterogeneous Reactions



A detailed gas chemistry model is applied to simulate the chemical evolution within the tagged column of air since it passed over the Salem power plant up to locations A, B, and C.

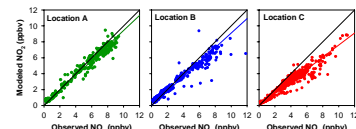
The objective is to examine the role and importance of the heterogeneous reaction of NO₃

Initial Conditions and Model Constraints:

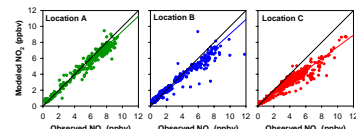
The model column points for each location are initialized and constrained as follows:
[O₃]₀ = 61 ppbv (corresponding to the uniform vertical O₃ profile at locations A and B)
[NO]₀ = observed NO_x at that location
Aerosol Size Dist. = observed PCASP Aerosol Size Dist. at that location (particles greater than 0.11 μm only)
T, P, RH = observed T, P, RH at that location
Model integration time (Δt) = (time at that location) – (time when column was over Salem power plant) (for location C, wind speed shear is taken into account when calculating Δt, and [O₃]₀ = 55 ppbv below 0.8 km, corresponding to the uniform vertical O₃ profile at location Z).

The predicted O₃ and NO₂ are compared with the observed values at the respective locations.

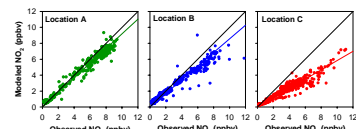
CASE 1a: $\gamma_{\text{N}_2\text{O}_5} = 0$, $\gamma_{\text{NO}_3} = 0$



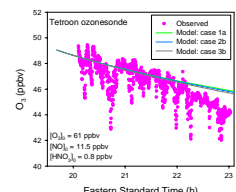
CASE 2a: $\gamma_{\text{N}_2\text{O}_5} = 0.001$, $\gamma_{\text{NO}_3} = 0$



CASE 3a: $\gamma_{\text{N}_2\text{O}_5} = 0.02$, $\gamma_{\text{NO}_3} = 0$

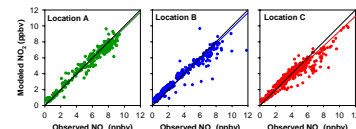


Model systematically under-predicts NO₂ as a function of processing time, especially at higher values of $\gamma_{\text{N}_2\text{O}_5}$. While the actual value of $\gamma_{\text{N}_2\text{O}_5}$ cannot be determined from this dataset, its value is thought to be between 0.001 to 0.02.

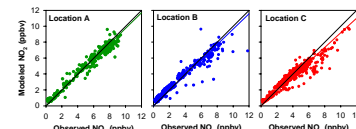


Modeled decay of O₃ agrees remarkably well with the Lagrangian measurement onboard the tetraon for the first two hours, but shows some deviation afterward.

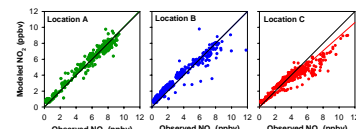
CASE 1b: $\gamma_{\text{N}_2\text{O}_5} = 0$, $\gamma_{\text{NO}_3} = 0.01$



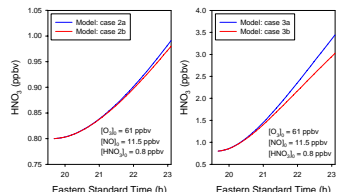
CASE 2b: $\gamma_{\text{N}_2\text{O}_5} = 0.001$, $\gamma_{\text{NO}_3} = 0.01$



CASE 3b: $\gamma_{\text{N}_2\text{O}_5} = 0.02$, $\gamma_{\text{NO}_3} = 0.04$



Including heterogeneous conversion of NO₃ to NO can largely correct the discrepancy at all three locations. Depending on the choice of $\gamma_{\text{N}_2\text{O}_5}$, the value of γ_{NO_3} ranges between 0.01 to 0.04, representing the upper limit.



The heterogeneous NO₃ to NO reaction tends to slow down HNO₃ production, and could be important at higher values of γ_{NO_3} .

7. CONCLUSIONS & CLOSING REMARKS

Model analysis of the quasi-Lagrangian aircraft dataset appears to provide indirect evidence of the heterogeneous conversion of NO₃ to NO (or possibly NO₂), with an uptake coefficient between 0.01 and 0.04. This represents an upper limit since the aerosol surface area is estimated using only particles greater than 0.11 μm diameter. Laboratory experiments are needed to study heterogeneous conversion of NO₃ on coal-fired power plant aerosols, which are composed of potentially reactive fly ash material.

8. ACKNOWLEDGEMENTS

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